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A Structural Study of 3'-Deoxytubercidin and 3-Deaza-3'-deoxyadenosine

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Abstract. 3'-Deoxytubercidin (3'-dTu), C₁₁H₁₄N₄O₃, $M_r = 250 \cdot 26$, monoclinic, $P2_1$, $a = 5 \cdot 299$ (1), b = 9.945 (3), $c = 11 \cdot 107$ (1) Å, $\beta = 100 \cdot 75$ (1)°, V =575.13 Å³, Z = 2, $D_x = 1.445$ Mg m⁻³, λ (Cu K α) = $1.54184 \text{ Å}, \mu = 86.2 \text{ mm}^{-1}, F(000) = 264, T = 298 \text{ K}.$ Final R = 0.030 for 1266 observed reflections with $I \ge 1.5\sigma(I)$. 3-Deaza-3'-deoxyadenosine (3'-ddA), C₁₁- $H_{14}N_4O_3$, $M_r = 250.26$, orthorhombic, $P2_12_12_1$, a =5.826 (1), b = 9.075 (2), c = 21.887 (3) Å, V =1157.27 Å³, Z = 4, $D_x = 1.436$ Mg m⁻³, λ (Cu Ka) = 1.54184 Å, $\mu = 85.6$ mm⁻¹, F(000) = 528, T = 298 K. Final R = 0.030 for 1075 observed reflections with $I \ge 1.5\sigma(I)$. 3-Deaza-3'-deoxyadenosine (3'-ddA) and 3'-deoxytubercidin (3'-dTu) are analogues of the nucleoside 3'-deoxyadenosine (3'-dA). The sugar pucker for both structures is the C3'-endo conformation. For 3'-dTu the C4'-C5 bond conformation is gauche(+)-trans whereas for 3'-ddA it is gauche(+)gauche(-). Where the substitutions of N by C have been made in the purine bases, the relevant bond lengths have increased by 0.05 Å and the bond angles by 3° compared with adenosine, which causes significant deviation from the usual planar purine-ring structure.

Introduction. The title structures are analogues of cordycepin, the nucleoside antitumour antibiotic 3'-deoxyadenosine (3'-dA) (Radwan & Wilson, 1980).

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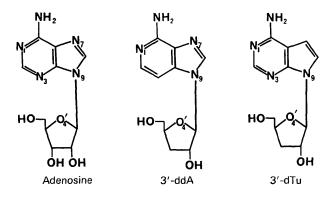
This has been shown to inhibit the growth of human tumour cells in culture (Rich, Meyers, Weinbaum, Cory & Suhadolnik, 1965). 3'-dA primarily inhibits RNA synthesis and, as a consequence, blocks DNA and protein synthesis in cells. Its mode of action involves incorporation into the 3'-end of an RNA molecule and this prevents further elongation, thereby acting as a chain terminator (Suhadolnik, 1979). A major problem associated with the clinical use of these nucleoside antitumour antibiotics is that the enzyme adenosine deaminase (ADAase) (Montgomery, 1970) causes deamination of purine antimetabolites and hence limits their activity. The aim of the structural studies detailed in this paper is to examine compounds potentially active in chain termination of RNA that are, at the same time, resistant to ADAase, although without being an inhibitor (Montgomery, Thomas, Zell, Einspahr & Bugg, 1985), which causes other toxic effects produced by excess adenosine (Plagemann & Wohlhueter, 1981). Syntheses are described elsewhere (Serafinowski, 1987).

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3-Deaza-3'-deoxyadenosine (3'-ddA) has a C atom substituted for the N atom at the 3-position of the adenine base; 3'-deoxytubercidin (3'-dTu) has a C atom substituted for an N atom at position 7. The substitutions of a C for an N atom in these nucleoside antibiotics have electronic effects which are manifested in changes in bond geometry. These are discussed in this paper and compared with the crystal structures of the parent nucleosides adenosine (Lai & Marsh, 1972), 2'-deoxyadenosine (Watson, Sutor & Tollin, 1965),

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3'-deoxyadenosine (Radwan & Wilson, 1980) and other related nucleosides.



Experimental. 3'-dTu. Recrystallization from ethanol/ water solution produced colourless rectangular crystals. A specimen of dimensions $0.2 \times 0.2 \times 0.1$ mm was used. Cell dimensions were obtained from least-squares refinement of 25 θ values measured on an Enraf– Nonius diffractometer; Ni-filtered Cu Ka radiation was used ($\lambda = 1.54178$ Å). Intensity data collected with an ω -2 θ scan technique and a max. scan time of 120 s per reflection, for $1.5 \le \theta \le 72^{\circ}$ and $0 \le h \le 6$, $0 \le k \le$ 12, $-13 \le l \le 13$, 1325 unique reflections were measured, of which 1266 had $I \ge 1.5\sigma(I)$.

3'-ddA. Recrystallization from ethanol/water solution produced colourless cubic crystals. A crystal of dimensions $0.2 \times 0.2 \times 0.2$ mm was used. Cell dimensions were obtained from least-squares refinement of 25 θ values measured on an Enraf-Nonius diffractometer; Ni-filtered Cu K α radiation was used. Intensity data collected with an ω -2 θ scan technique and a max. scan time of 120 s per reflection, for $1.5 \le \theta \le 70^{\circ}$ and $0 \le h \le 7$, $0 \le k \le 11$, $0 \le l \le 26$, 1299 unique reflections, of which 1075 had $I \ge 1.5\sigma(I)$.

Three intensity standards were monitored every 200 reflections during the data collection for both structures and showed no statistically significant crystal decay. An empirical absorption correction was applied to both data sets (Walker & Stuart, 1983). The structures were solved by direct methods with MULTAN82 (Main et al., 1982). H atoms were located in difference Fourier syntheses, and their positional and isotropic thermal parameters refined on F by full-matrix least-squares methods, together with non-H-atom positional and anisotropic thermal parameters. For 3'-dTu, the final R was 0.030 and wR was 0.028, for 3'-ddA, the final R was 0.030 and wR 0.029. For both structures, unit weights were found to be appropriate. Scattering factors were taken from International Tables for X-ray Crystallography (1974). Calculations were performed on a VAX 11/750 computer using the SDP system (Frenz, 1980). For both 3'-dTu and 3'-ddA, max. Δ/σ was 0.01, and observed range of $\Delta \rho$ was within $\pm 0.15 \text{ e} \text{ Å}^{-3}$.

Discussion. The molecular structures of 3'-dTu and 3'-ddA are shown in Fig. 1. Atomic parameters and bond distances and angles are given in Tables 1 and 2.* The adenine-base analogues show significant deviations from the usual planar purine-ring structures (Table 2), this being most prominent for 3'-dTu. For this structure the C7 and N1 atoms are 0.05 Å below the mean plane and the N6 atom is 0.06 Å above. With 3'-ddA the C3 and N6 atoms are 0.04 Å above the mean plane and N1 is 0.04 Å below it. These deviations are significantly greater than those for other modified nucleosides, such as tubercidin (Stroud, 1973) and togacamycin mono-hydrate (Prusiner & Sundaralingam, 1978).

* Lists of H-atom parameters, anisotropic thermal parameters, H-bond distances and structure factors have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 44034 (20 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

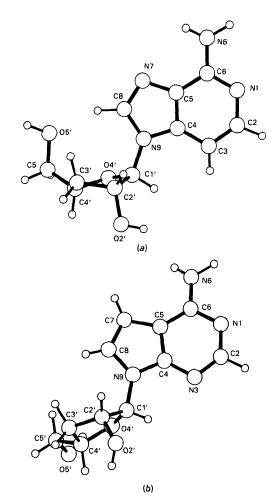


Fig. 1. Molecular structures of (a) 3-deaza-3'-deoxyadenosine, (b) 3'-deoxytubercidin.

The bond lengths and angles for both structures are in good agreement with those of adenosine, apart from where the substitutions have been made (Table 2). The overall effect is that the relevant bond lengths in both structures have increased by 0.05 Å and the bond angles by 3° compared with adenosine.

In both structures the sugar pucker may generally be described as being in the C3'-endo conformational range. The pseudorotational phase angle P for 3'-dTu is 25.7 (9)° and for 3'-ddA it is 7.7 (4)° (Altona & Sundaralingam, 1972).

The glycosidic angle for 3'-dTu is -106.4 (2)° and
for 3'-ddA it is $-126 \cdot 1$ (2)°. Thus they are both in an
anti conformation (Fig. 1 and Table 3). This is in
accord with the majority of nucleosides in the solid
state. It can be seen that both structures have values for
the glycosidic angles similar to tubercidin (Altona &

Table 2. Bond distances (Å), bond angles (°) and deviations (Å) of atoms from least-squares planes

Atomic names in parentheses signify where a substitution has been made relative to adenosine.

Tabla	1 Non II a	tom moniti		adama	3'-dTu 3'-ddA 3'-dTu 3'-ddA						
Table	1. Non-H-a			elers, with	NI C2 1-339 (6) 1-356 (6) C2 NI C6 118-0 (4) 119-1 (4)						
	e.s.	d.'s in parer	<i>itheses</i>		N1 C6 1.348 (6) 1.339 (6) N1 C2 N3 (C3) 129-0 (5) 125-8 (4)						
		•			C2 N3 (C3) 1·316 (6) 1·373 (6) C2 N3 (C3) C4 111·4 (4) 114·6 (4)						
	B is in $Å^2$ and	is defined as 4	$(B_{11} + B_{22} + B_{23})$)	N3 (C3) C4 1-353 (5) 1-394 (6) N3 (C3) C4 C5 126-7 (4) 121-5 (4)						
	<i>B</i> is in A^2 and is defined as $\frac{1}{3}(B_{11} + B_{22} + B_{33})$.				C4 C5 1·391 (6) 1·382 (6) N3 (C3) C4 N9 125·1 (4) 132·9 (4) C4 N9 1·367 (5) 1·381 (5) C5 C4 N9 108·3 (4) 105·6 (4)						
	x	у	Z	B (Å ²)	C5 $C6$ 1.404 (6) 1.411 (6) $C4$ $C5$ $C6$ 115.3 (4) 119.8 (4)						
3'-dTu		•		. ,	C5 C7 (N7) 1.432 (7) 1.388 (5) C4 C5 C7 (N7) 107.4 (4) 110.6 (4)						
N1	0.1100 (4)	0.037	0.1943 (2)	3.22 (4)	C6 N6 1.346 (6) 1.352 (5) C6 C5 C7 (N7) 137.2 (4) 129.7 (4)						
C2	0.0982 (5)	0.0205 (3)	0.0842 (2)	3.06 (5)	C7 (N7) C8 1-349 (6) 1-312 (5) N1 C6 C5 119-3 (4) 119-2 (4)						
N3	-0.0622 (3)	0.0037 (2)	0.0190 (2)	2.76 (4)	C8 N9 1-393 (6) 1-370 (5) N1 C6 N6 117-1 (4) 119-3 (4)						
C4	-0.2411(4)	0.0958 (2)	0.0031 (2)	2.35 (4)	N9 C1' 1.447 (5) 1.464 (5) C5 C6 N6 123.6 (5) 121.5 (4)						
C5	-0.2575 (4)	0.1633 (2)	0.1049 (2)	2.51 (4)	C1' C2' 1.538 (6) 1.535 (5) C5 C7 (N7) C8 106.4 (4) 104.4 (4)						
C6	-0.0621 (5)	0.1338 (2)	0.2053 (2)	2.86 (4)	C1' O4' 1-419 (5) 1-402 (4) C7 (N7) C8 N9 110-2 (4) 113-3 (3)						
N6	-0.0371	0.1961 (3)	0.3145 (2)	4.28 (5)	C2' C3' 1.505 (7) 1.518 (5) C4 N9 C8 107.8 (4) 106.2 (4)						
C7	-0.4777 (5)	0.2494 (3)	0.0796 (2)	2.83 (4)	C2' C2' 1.433 (6) 1.423 (5) C4 N9 C1' 124.8 (4) 125.3 (4)						
C8	-0.5813 (4)	0.2317 (3)	-0.0398 (2)	2.84 (4)	C3' C4' 1.524 (6) 1.510 (7) C8 N9 C1' 126.6 (4) 128.4 (3)						
N9	-0-4394 (3)	0.1377 (2)	-0·0921 (2)	2.53 (4)	C4' C4' 1.432 (5) 1.451 (6) N9 C1' C2' 115.1 (4) 114.2 (3)						
C1′	-0.4654 (4)	0.1068 (3)	-0.2211 (2)	2.67 (4)	C4' C5' 1-498 (7) 1-499 (6) N9 C1' O4' 108-3 (4) 109-3 (3)						
C2'	-0.7340 (5)	0.0595 (3)	-0.2841 (2)	3.20 (5)	C5' O5' 1.421 (6) 1.426 (5) C2' C1' O4' 107.1 (4) 106.9 (3)						
C3'	-0.8434 (5)	0.1804 (3)	-0.3572 (2)	3.42 (5)	C1' C2' C3' 103-4 (4) 102-2 (3)						
C4'	-0.6083 (4)	0.2526 (3)	-0.3861 (2)	2.63 (4)	C1' C2' O2' 110-3 (4) 109-8 (3)						
O4' O2'	-0.4088 (4)	0.2247 (2)	-0.2831 (1)	3-56 (3)	C3' C2' O2' 113-3 (4) 109-2 (3)						
C5'	-0·7142 (4) -0·6387 (5)	-0.0565 (2) 0.4020 (3)	0·3583 (2) 0·3989 (2)	4.45 (4)	C2' C3' C4' 104·0 (4) 103·5 (4) C3' C4' O4' 104·5 (4) 104·9 (4)						
05'	-0.4104 (4)	0.4679 (2)	-0.3989 (2) -0.4165 (1)	3-80 (6) 4-01 (4)	C3' C4' O4' 104-5 (4) 104-9 (4) C3' C4' C5' 114-2 (5) 116-9 (5)						
05	-0-4104 (4)	0.4079 (2)	-0.4103 (1)	4.01 (4)	O4' C4' C5' 108-6 (4) 108-7 (3)						
3'-ddA					C1' C4' C4' C4' 110.9 (4) 111.0 (3)						
C2	0.6111 (5)	0.4395 (3)	0-4785 (1)	2.59 (5)	C4' $C5'$ $O5'$ $113.0 (5) 111.3 (4)$						
Č6	0.9350 (5)	0.4535 (3)	0.4169(1)	2.39 (5)							
04'	0.4058 (3)	-0.0264 (2)	0.32519 (8)	2.54 (3)	Deviations from least-squares planes: e.s.d.'s ca 0.002 Å; the atoms marked						
C3	0.5372 (5)	0.3063 (3)	0.4563 (1)	2.34 (5)	with an asterisk were used to calculate the mean planes						
N1	0-8040 (4)	0.5117 (2)	0-46106 (9)	2.49 (4)	•						
C4	0.6805 (5)	0.2437 (3)	0.4123(1)	2.10 (5)	3'-dTu						
N7	0-9882 (4)	0.2311 (2)	0.3484 (1)	2.86 (5)	Purine ring; $\chi^2 = 3328$						
N6	1-1188 (4)	0.5299 (3)	0.3964 (1)	3-28 (5)	N1* -0.054 C5* -0.018 C8* -0.11						
N9	0-6727 (4)	0.1127 (2)	0.38029 (9)	2.17 (4)	C2* -0.030 C6* 9.065 N9* 0.028						
C5	0.8762 (5)	0.3149 (3)	0-3923 (1)	2.13 (5)	N3* 0-041 N6* 0-064 C1' 0-273 C4* 0-020 C7* -0-045						
C3'	0.6129 (6)	-0.2360 (3)	0.3554 (1)	2.89 (5)	C4* 0.020 C7* -0.045						
C8	0-8614 (5)	0.1124(3)	0.3429 (1)	2.84 (5)	3'-ddA						
C4' C2'	0.4423 (5)	-0.1797 (3)	0-3090 (1)	2.50 (5)	Purine ring; $\chi^2 = 1051$						
C2' C5'	0.5617(5)	-0.1445(3)	0.4118(1)	2.51 (5)							
02'	0·5119 (7) 0·3743 (4)	-0·1877 (4) -0·2086 (2)	0.2432(1)	3.95 (7)	N1* -0.047 C5* -0.004 C8* -0.012						
C1'	0.3743 (4)	0.0035 (3)	0-44395 (8)	3·29 (4)	$C2^{*}$ -0.005 $C6^{*}$ -0.008 $N9^{*}$ 0.002						
05'	0.4883 (3)	-0.1168(3)	0·3841 (1) 0·2331 (1)	2·17 (5) 4·74 (5)	C3* 0-033 N6* 0-043 C1' 0-0080 C4* 0-011 N7* -0-014						
	0.1212(3)	-0.1100 (3)	0.2331 (1)	4.14 (2)	C4* 0.011 N7* -0.014						

Table 3. Comparison of torsion angles (°) for modified nucleosides

04'-C1'-N9-C4 05'-C5'-C4'-04'	3-dTu 253-6 59-3	3-dA 233-9 64-3	A 188-6 60-1	2'-dA 182-7 68-1	araA 204-8 70-5	3'-dA 205∙6 62∙4	3'-dTu 246-5 61-9	3'-ddA 191-8 201-8			
O5'-C5'-C4'-C3'	175.5	305-8	176.9	186.9	55-1	180.0	180.0	201·8 60·5			
Sugar pucker	C3'-endo	C3'-endo	C3'-endo	C3'-exo	C3'-endo	C3'-endo	C3-endo	C3'-endo			
	$3'-dTu \equiv $ $3'-ddA \equiv $ $2'-dA$ $araA$ $3'-dA$ $3'-dA$ $3-dTu$	This study. Adenosine (Lai & Marsh, 1972). 2'-Deoxyadenosine (Watson, Sutor & Tollin, 1965). Adenine arabinofuranoside (Chwang & Sundaralingam, 1974). 3'-Deoxyadenosine (Radwan & Wilson, 1980). Tubercidin (Altona & Sundaralingam, 1972).									

3-dA 3-Deazaadenosine (Singh, May, Townsend & Hodgson, 1976).

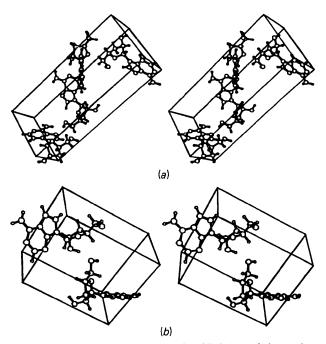


Fig. 2. Unit-cell packing diagrams for (a) 3-deaza-3-deoxyadenosine, (b) 3'-deoxytubercidin.

Sundaralingam, 1972), which is somewhat higher than for the other related nucleosides. For the majority of modified nucleosides listed in Table 3, including 3'-dTu, the conformation about the C4'-C5' bond is gauche(+)-trans. 3'-ddA is exceptional here in having a gauche(+)-gauche(-) conformation about this bond, which is normally the more stable conformation for nucleosides.

Hydrogen bonding

Fig. 2 shows stereoviews of the packing diagrams of the molecules in their crystal lattice. Hydrogen-bond

lengths and angles have been deposited. There is a lack of base-base hydrogen bonds in both crystal lattices as compared with most nucleosides in the solid state.

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Structure of 2-Methoxy-N-(4-nitrobenzylidene)-5-pyridylamine

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Abstract. $C_{13}H_{11}N_{3}O_{3}$, $M_r = 257 \cdot 2$, monoclinic, $P2_1/n$, $a = 11 \cdot 080$ (2), $b = 3 \cdot 376$ (1), $c = 14 \cdot 364$ (2) Å, $\beta = 112 \cdot 56$ (1)°, $V = 1231 \cdot 1$ (6) Å³, Z = 4, $D_x = 1 \cdot 388$, $D_m = 1 \cdot 385 \text{ g cm}^{-3}$, $\lambda (Mo K\alpha) = 0.71069$ Å, $\mu = 0.62 \text{ cm}^{-1}$, F(000) = 536, T = 293 K, R = 0.0374 for 1554 observed reflections. The molecule as a whole is in a nearly planar conformation. The dihedral angle between the plane through C(1), C(7), N(1) and C(8) and the phenyl ring is $3 \cdot 2$ (2)°, and that between the phenyl and pyridine rings is $19 \cdot 6$ (2)°. Bond distances indicate a quinoid resonance contribution throughout the molecule.

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